

# Study of mass attenuation coefficients, effective atomic numbers and electron densities of carbon steel and stainless steels

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**ABSTRACT** The mass attenuation coefficient ( $\mu/\rho$ ), effective atomic number ( $Z_{\text{peff}}$ ) and effective electron density ( $N_{\text{eff}}$ ) of carbon steel and stainless steels were computed by using the WinXcom program. The  $\mu/\rho$  values are higher in the photo-electric absorption and pair production region and approximately constant in the Compton scattering region. The variation in  $\mu/\rho$  values is explained by partial interaction processes. The effective atomic numbers of the steels were also calculated by the Auto- $Z_{\text{eff}}$  program. The  $Z_{\text{peff}}$  values of the steels were compared with experimental results for available energies and were found to be in good agreement. SS304L shows equal values of effective atomic number by both the methods in the region of interest. The electron densities of the selected steels are observed to be equal and approximately constant in the Compton scattering region.

**Keywords:** steel / effective atomic number / mass attenuation coefficient / Auto- $Z_{\text{eff}}$

## 1. Introduction

The application of gamma radiation has been increased in science and technology; industry, medicine, agriculture, petroleum plants, energy sectors, etc. In practice, gamma-ray sources above 200 keV up to 1500 keV are widely used in radiography, archaeometry, chemotherapy, medical diagnosis and Compton scatter images (Hubbel *et al.*, 1975). Gamma rays can be absorbed in high-density materials such as lead, tungsten, concretes and building materials. Gamma-ray interaction is dependent upon the photon energy and atomic number of the elements of the composite materials. The absorption of gamma rays is represented by the mass attenuation coefficients, molecular cross-sections, atomic cross-sections, effective atomic numbers and electron densities. The linear attenuation coefficient for a material depends on the incident photon energy, atomic number and density. The attenuation coefficient is defined as the probability of radiation interacting with a material per unit path length (Woods, 1982). Hine (1952)

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pointed out that in the study of gamma interaction with complex media, the atomic number of the complex media cannot be represented by a single value for the entire energy, similar to the elements. This is due to the fact that gamma-ray interaction is dependent on the atomic number. Thus, a term called the “effective atomic number” should be defined for such complex materials. The effective atomic number is a convenient parameter for evaluation of gamma-ray interaction with a material. The effective atomic number can provide an initial estimation of the chemical compositions of the complex, neutron/gamma-ray shielding effectiveness and other parameters. In general, the effective atomic number of inorganic compounds, metals and alloys is large (Singh and Badiger, 2012a, 2012b), while it is small for organic substances (Singh and Badiger, 2013). The other important quantity for interaction is the effective electron number or electron density, as it is defined as the electrons per unit mass of the absorber.

The mass attenuation coefficient, effective atomic number and electron density are the gamma-ray interaction parameters to describe shielding effectiveness. Mass attenuation coefficients of elements are taken from standard tables (Berger and Hubbell, 1987/99) developed as the XCOM program for theoretical values for mass attenuation coefficients and cross-sections for various elements, compounds and mixtures at energies of 1 keV to 100 GeV. The XCOM program was converted to the Windows platform and is known as WinXcom (Gerward *et al.*, 2001, 2004), which is very useful for analysis.

Gamma-ray absorption is a vital area; various researchers have carried out studies for different types of materials. Effective atomic numbers of spin ice compounds, gaseous mixtures, alcohols, alloys, solid-state neutron track detectors, vitamins and composite mixtures have been obtained (İçelli *et al.*, 2005; Bastug *et al.*, 2010; Medhat, 2011; Demir *et al.*, 2012; Singh and Badiger, 2012a, 2012b, 2013). Mass attenuation coefficients, effective atomic numbers and electron densities of thermoluminescent dosimetric compounds (Shivaliga *et al.*, 2004) have also been obtained. The mass attenuation coefficients, and effective atomic and electron numbers for some alloys such as Cr, Fe and Ni at different energies have been reported (Kaewkhao *et al.*, 2008; Han *et al.*, 2009). The effective atomic numbers of various alloys have been estimated (Prasad *et al.*, 1998; El-Kateb *et al.*, 2000; Murty *et al.*, 2000; İçelli *et al.*, 2005; Akkurt, 2007, 2009).

In the present work, we computed the mass attenuation coefficients, effective atomic numbers and electron densities of the carbon steel and stainless steels given in Table I. The compositions of the stainless steels (1 to 4) are taken from the proceeding (Prasanna and Kumar, 2012) as supplied by the manufacturer. The theoretical values of the effective atomic numbers of these materials were

**TABLE I**  
**Chemical composition of carbon steel and stainless steels.**

	Composition (weight percentage)											
	C	Al	Si	P	S	Ti	Cr	Mn	Fe	Ni	Cu	Mo
Steel 1	0.4200	0.0490	0.2500	0.0210	0.1100	0.0019	1.0500	0.8900	97.0371	0.0600	0.0100	0.2000
Steel 2	0.3900	0.0360	0.2600	0.0140	0.0080	0.0014	0.0400	0.7900	98.4206	0.0200	0.0100	0.0100
Steel 3	1.0300	0.0340	0.3100	0.0150	0.0120	0.0026	1.4300	0.3200	96.7960	0.0300	0.0100	0.0100
Steel 4	0.0330	0.0400	0.2600	0.0200	0.0210	0.0013	0.8900	0.6600	98.5777	0.0300	0.0100	0.1600
CS516	0.3100	0.0000	0.1900	0.0100	0.0050	0.0000	0.0000	0.9800	98.3900	0.0000	0.0000	0.0000
SS410	0.1500	0.0000	1.0000	0.0400	0.0300	0.0000	11.2500	1.0000	85.7800	0.7500	0.0000	0.0000
SS316L	0.0300	0.0000	0.5300	0.0400	0.0050	0.0000	16.0000	1.8500	68.2700	10.7700	0.0000	2.1200
SS304L	0.0000	0.0000	0.0000	0.0000	0.0000	0.0000	18.0000	0.0000	74.0000	8.0000	0.0000	0.0000

compared with available experimental values at 279.1 keV, 661.6 keV, 662 keV, 1115.5 keV, 1173 keV and 1332 keV, which show good agreement.

## 2. Computational work

Computation of the mass attenuation coefficients of the carbon steel and stainless steels was carried out by the mixture rule by using the WinXcom program for the energies 1 keV to 100 GeV (Gerward *et al.*, 2001, 2004). Computation of the effective atomic numbers and electron densities of carbon steel and stainless steel by total gamma photon interaction was carried out by a practical formula (Manohara *et al.*, 2008). The atomic numbers and atomic masses of the elements were taken from the atomic weights of the elements 2009, IUPAC (Michael and Tyler, 2011). The steel samples, steel 1, steel 4, CS516 and SS316L, are unnormalized compositions of elements which have been normalized to 1 by the software for calculation. As the difference is insignificant, the impact on the results will be negligible.

### 2.1. Mass attenuation coefficients

The mass attenuation coefficient  $(\mu/\rho)_{\text{steel}}$  is given by:

$$(\mu/\rho)_{\text{steel}} = \sum_i^n w_i \left( \frac{\mu}{\rho} \right)_i \quad (1)$$

where  $w_i$  is the proportion by weight and  $(\mu/\rho)_i$  is the mass attenuation coefficient of the  $i$ th element.

The quantity  $w_i$  is given by:

$$w_i = \frac{n_i A_i}{\sum_j n_j A_j} \quad (2)$$

with the condition that

$$w_1 + w_2 + w_3 + \dots + w_i = 1 \quad (3)$$

where  $A_i$  is the atomic weight of the  $i$ th element and  $n_i$  is the number of formula units in the compounds.

### 2.2. Effective atomic numbers

The effective atomic number ( $Z_{\text{Pleff}}$ ) for total photon interaction is given by:

$$Z_{\text{Pleff}} = \frac{\sum_i f_i A_i \left(\frac{\mu}{\rho}\right)_i}{\sum_j f_j \frac{A_j}{Z_j} \left(\frac{\mu}{\rho}\right)_j} \quad (4)$$

where  $f_i$  is the molar fraction in the mixture/alloys,  $\mu$  is the linear attenuation coefficient,  $\rho$  is density,  $\mu/\rho$  is the mass attenuation coefficient,  $A$  is the atomic weight,  $Z$  is the atomic number, and the ratio  $A/Z$  between the atomic mass and the atomic number is approximately constant. Recently, a new software program has made the tedious calculations of the present work very easy. This software Auto- $Z_{\text{eff}}$  is freely available and the effective atomic numbers of soft tissue, brass, oleic acid,  $\text{CaSO}_4$ : Dy, water and alloys have been estimated (Taylor *et al.*, 2012). The effective atomic numbers calculated by Auto- $Z_{\text{eff}}$  and experimental values demonstrated good agreement.

### 2.3. Electron densities

The electron density is given by  $N_{\text{eff}} = N_A Z/A$ , which is generalized as:

$$N_{\text{eff}} = N_A \frac{n Z_{\text{Pleff}}}{\sum_i n_i A_i} = N_A \frac{Z_{\text{Pleff}}}{\langle A \rangle} \quad (5)$$

where  $n_i$  is the number of atoms of the  $i$ th constituent element,  $n$  is the total number of atoms and  $\langle A \rangle$  is the average atomic mass of the steel alloys.

### 3. Results and discussion

Variation in mass attenuation coefficients for the selected alloys for the gamma-ray energy range 1 keV to 100 GeV is shown in Figure 1. Partial photon interaction with these steels is shown in Figure 2 (a–h) for understanding of the interaction processes (photo-electric absorption, Compton scattering and pair production) and their contribution. The effective atomic numbers and electron densities of the selected steels are shown in Figures 3 and 4, respectively, for the photon energies 10 keV to 1 GeV. Gamma-ray interaction is based on photo-electric absorption, Compton scattering and pair production. Therefore, we studied separately the mass attenuation coefficients of steels for coherent, incoherent, photo-electric absorption and pair (nuclear/electron).

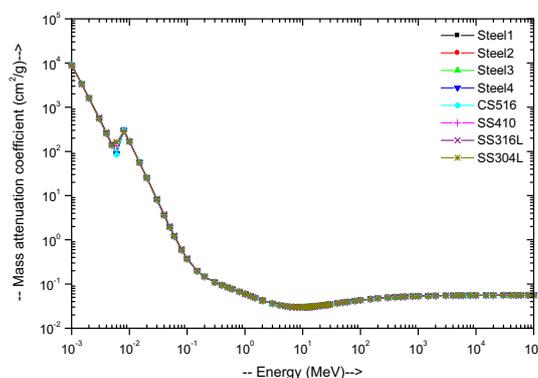
#### 3.1. Mass attenuation coefficients

Variation in  $(\mu/\rho)_{\text{steel}}$  with gamma photon energy is shown in Figure 1. From Figure 1, it is clear that the  $(\mu/\rho)_{\text{steel}}$  of the steels decreases with an increase in energy of the incident photon. It can be seen that the  $(\mu/\rho)_{\text{steel}}$  values decrease sharply up to 100 keV photon energy, further decrease very slowly in the photon energy region 100 keV to 10 MeV, and afterward increase again and become constant. The  $(\mu/\rho)_{\text{steel}}$  values of all the steels show one sharp peak at 7.11 keV. This peak is due to k-edge absorption for nickel and copper, which lies at the photon energies 8.3 keV and 8.979 keV, respectively. At around 1 MeV photon energy, the  $(\mu/\rho)_{\text{steel}}$  values are found to be constant. Complete variation in  $(\mu/\rho)_{\text{steel}}$  with photon energy for partial interaction processes is explained in the next sub-sections.

##### 3.1.1. Photo-electric absorption process

From Figure 2 (a–h), it is observed that the values of  $\mu_{\text{steel, photo}}$  decrease rapidly with an increase in incident photon energy for all the selected steels. Initially, at 1 keV, the value of  $\mu_{\text{m (photo)}}$  is very high; of the order of  $10^3$ , which decreases sharply. The values of  $\mu_{\text{steel, photo}}$  of all the steels become negligible ( $<10^{-3}$ ) as photon energy reaches 100 keV. The reason for such variation may be due to dependency of the photo-electric cross-section on photon energy as  $E^{3.5}$ . Hence, the  $\mu_{\text{m (photo)}}$  values decrease rapidly at low energy. This shows the dominance of photo-electric absorption in the low photon energy region.

It can be observed that there is slight variation in the  $\mu_{\text{steel, photo}}$  of all the selected steels at particular photon energies. It is due to the reason that photo-electric absorption is dependent on the atomic numbers of the elements of the mixture or compounds as  $Z^{4.5}$ . The steels in our investigation contain different



**Figure 1 – Variation in mass attenuation coefficients of carbon steel and stainless steels with photon energy.**

constituent elements ( ${}^6\text{C}$ ,  ${}^{13}\text{Al}$ ,  ${}^{14}\text{Si}$ ,  ${}^{15}\text{P}$ ,  ${}^{16}\text{S}$ ,  ${}^{22}\text{Ti}$ ,  ${}^{24}\text{Cr}$ ,  ${}^{25}\text{Mn}$ ,  ${}^{26}\text{Fe}$ ,  ${}^{28}\text{Ni}$ ,  ${}^{29}\text{Cu}$  and  ${}^{42}\text{Mo}$ ) with different weight fractions or numbers of atoms, as given in Table I. The slight difference in the weight fraction of elements results in insignificant variation in  $\mu_{\text{m}}(\text{photo})$  values.

### 3.1.2. Compton scattering process

From Figure 2 (a–h), it is observed that  $\mu_{\text{steel, coherent}}$  decreases from 1 keV very fast and becomes insignificant at around 1 MeV, whereas  $\mu_{\text{steel, coherent}}$  increases from 1 keV to 100 keV, thereafter decreases, and becomes insignificant at around 200 MeV photon energy. For the combined effect of Compton scattering, it is observed that  $\mu_{\text{steel, coherent}}$  initially increases from 1 keV to 10 keV, thereafter becomes constant up to 100 keV and, with a further increase in the incident photon energy,  $\mu_{\text{steel, coherent}}$  decreases slowly up to 200 MeV. Further, it is also observed that  $\mu_{\text{steel, coherent}}$  and  $\mu_{\text{steel, photo}}$  values are equivalent at 20 keV. Above 20 keV photon energy, it can be seen that Compton scattering is the dominant process of interaction up to 20 MeV for all the selected steels. It can also be noted that the variation in  $\mu_{\text{steel, coherent}}$  is slower compared with  $\mu_{\text{steel, photo}}$ .

### 3.1.3. Pair production process

From Figure 2 (a–h), it can be observed that the mass attenuation coefficient values for the pair production process in the nuclear field,  $\mu_{\text{steel, npp}}$ , and electron field,  $\mu_{\text{steel, epp}}$ , start at the threshold energies of 1.022 MeV and 2.044 MeV and thereafter increase with an increase in the incident photon energy. In this high-energy region, a slight variation is again seen in  $\mu_{\text{steel, pp}}$  due to their different chemical compositions at particular incident photon energies (Fig. 2 (a–h)). It may

PHOTON INTERACTION WITH CARBON STEEL AND STAINLESS STEELS

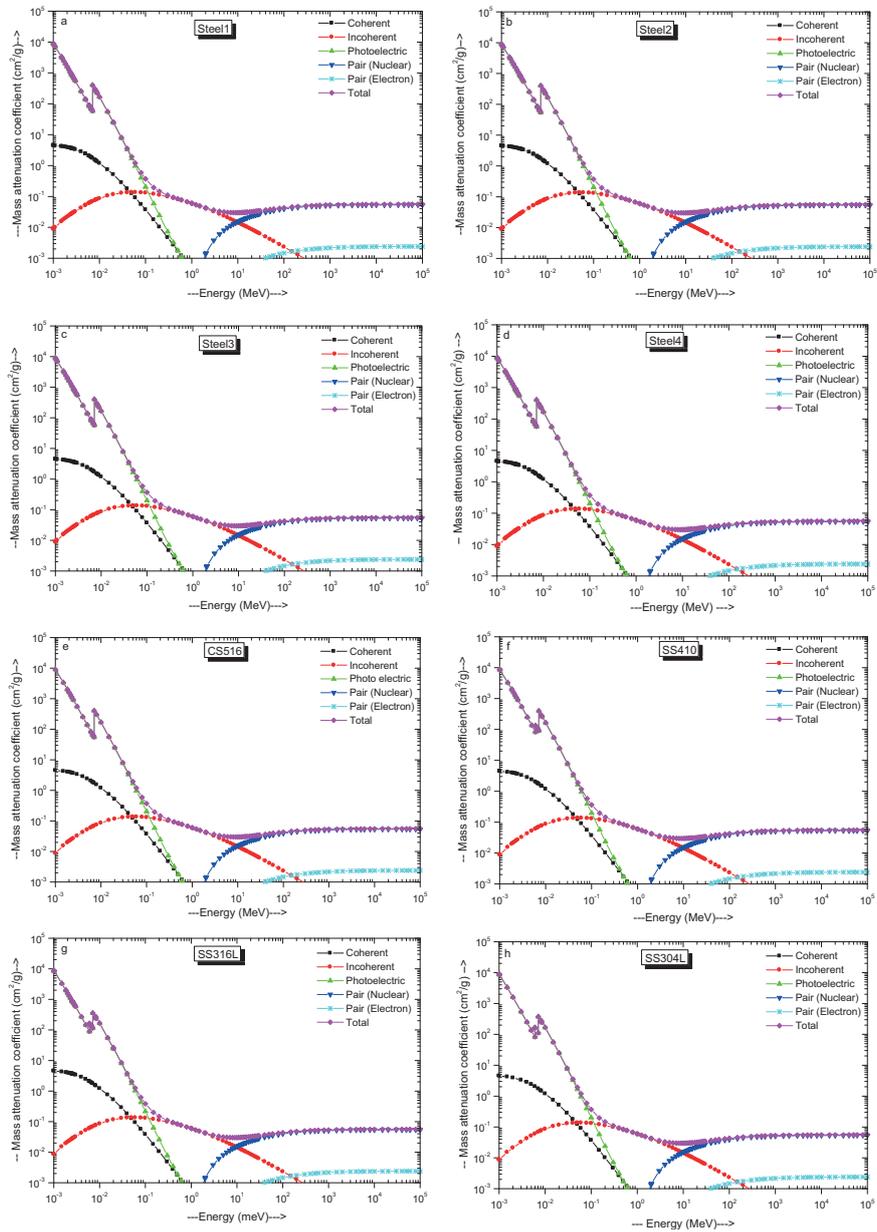
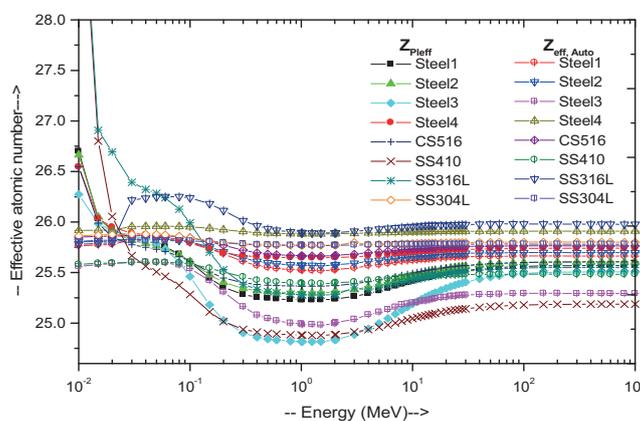


Figure 2 (a-h) – Variation in mass attenuation coefficients for partial photon interactions viz. photoelectric, Compton and pair production for carbon steel and stainless steels.



**Figure 3 – Variation in effective atomic numbers of carbon steel and stainless steels with photon energy for the practical formula and robust method.**

be due to the fact that the cross-section of the pair production process, in the nuclear field, is  $Z^2$ -dependent, whereas for triplet production, the cross-section varies linearly with  $Z$ . It is also observed from Figure 2 (a–h) that although the pair production process initiates from 1.022 MeV, it requires more incident photon energy to overcome Compton scattering and to become the dominant photon interaction process. For the all selected steels,  $\mu_{\text{steel, pp}}$  becomes equivalent to  $\mu_{\text{steel, Compton}}$  at about 20 MeV photon energy.

Finally, it is concluded that the Compton scattering dominant region is from 100 keV to 20 MeV where  $\mu_{\text{steel, Compton}}$  values are almost constant within an error of 10% due to linear dependency on  $Z$ . At low photon energy (<100 keV) the variation is found to be very large in the photon energy and atomic number of constituent elements.

### 3.2. Effective atomic numbers

Variation in  $Z_{\text{pleff}}$  for total photon interaction with incident photon energy is shown in Figure 3. The variation in  $Z_{\text{pleff}}$  of the steels can be explained by the photo-electric absorption, Compton scattering and pair production process discussed earlier. In the photon energy range 10 keV to 100 KeV,  $Z_{\text{pleff}}$  values of all the steels are found to be approximately constant, whereas a very large variation in  $Z_{\text{pleff}}$  values is observed. As a photon crosses photo-absorption and reaches the Compton region the  $Z_{\text{pleff}}$  value becomes constant. Above 20 MeV photon energy, the  $Z_{\text{pleff}}$  starts increasing and becomes invariable at 100 MeV energy. It can be

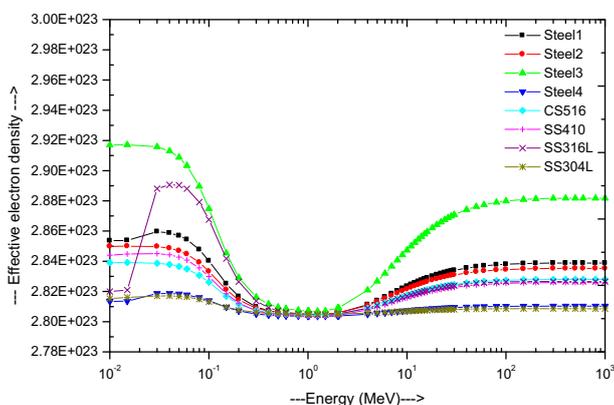


Figure 4 – Variation in effective electron densities of carbon steel and stainless steels with photon energy.

observed that the variation in  $Z_{\text{pleff}}$  in the Compton region is very narrow, i.e. 24.8 to 25.8, which is due to a small variation in the weight fraction of the constituent elements. The maximum value of  $Z_{\text{pleff}}$  for SS304L was in the Compton scattering process and pair-production regions, whereas it was minimum for Steel 3. The variation in  $Z_{\text{pleff}}$  values can be explained by the contribution of elements in the steel (which is  $Z$ -dependent).

A high value of  $Z_{\text{pleff}}$  of steels is expected below 10 keV photon energy due to absorption at the  $k$ -edge of the constituent elements. Except for the SS304L steel, all the others have Al, Si, P and S elements, for which  $k$ -edge absorption is at 1.56 keV, 1.839 keV, 2.145 keV and 2.472 keV. In the case of SS304L steel,  $k$ -edge absorption appears at 5.989 keV, 7.112 keV and 8.333 keV for Cr, Fe and Ni.

The effective atomic numbers calculated by  $Z_{\text{pleff}}$  and the Auto- $Z_{\text{eff}}$  program are shown in Figure 3. From Figure 3, it can be seen that there is small variation in the estimations in the Compton scattering region, whereas there is large variation in the photo-electric absorption region. The  $Z_{\text{pleff}}$  values are higher than those of  $Z_{\text{eff, auto}}$  in the photo-electric region, whereas there is the reverse behavior in the Compton region. SS304L shows equal values of the effective atomic number for the entire photon energy region 10 keV to 1GeV. The  $Z_{\text{pleff}}$  values of SS410 and SS316L are found to be very high in the photo-electric absorption region.

### 3.3. Electron densities

Figure 4 shows the variation in  $N_{\text{eff}}$  with incident photon energy for all the selected steels. Among the selected steels, steel 1 shows the maximum value for electron

density, whereas the minimum value for electron density was observed for SS304L in the 1 keV to 1 GeV energy region. The variation in electron density with incident photon energy follows the reciprocal trend, as observed for the total photon interaction effective atomic number. Under the dominance of any photon interaction, electron density remains constant for particular steels. However, a large variation in electron density is observed when the interaction process changes.

### **3.4. Comparison with experimental results**

The experimental investigations of the effective atomic numbers of the selected steels were carried out for some photon energies in the Compton scattering region. The effective atomic numbers for SS316L and CS516 stainless steels were experimentally investigated at 662 keV, 1173 keV and 1132 keV photon energies by  $^{137}\text{Cs}$  and  $^{60}\text{Co}$  sources (Fakarudin *et al.*, 2011). Differential incoherent scattering cross-sections and the effective atomic numbers of steel alloys were carried out in the energy region 279.1 keV to 1115.5 keV by  $^{203}\text{Hg}$ ,  $^{65}\text{Zn}$  and  $^{137}\text{Cs}$  sources at scattering angles of  $60^\circ$ ,  $80^\circ$  and  $100^\circ$  (Prasanna and Kumar, 2012). It is found that the steel samples 1 to 4 are characterized by a mean effective atomic number of 25.85 for Compton scattering. It is also found that the variation in effective atomic numbers for  $60^\circ$ ,  $80^\circ$  and  $100^\circ$  is insignificant as well as almost independent of incident photon energy. Slight differences in experimental effective atomic numbers and the present study may be due to non-consideration of the molecular, chemical and crystalline environment of atoms in the mixture rule. The experimental values of effective atomic numbers were compared with the  $Z_{\text{pleff}}$  of the present study and were found to be in good agreement, as given in Table II.

### **3.5. Uncertainties in results**

The mass attenuation coefficients of the steels were evaluated by the weighted sum of the mass attenuation coefficients of the elements. However, the molecular, chemical or crystalline environment of the atom changes the atomic wave function. This change in wave function limits the mixture rule applications. With the exception of fine-structure regions above the absorption edge, for photon energies above 10 keV, errors from these sources are expected to be less than a few percent. In the Compton region (10 keV to 1 MeV) the uncertainty in mass attenuation coefficients is expected to be less than 1 to 2% because of partitioning of experimental totals into photo effect and scattering totals. At medium energies (1 to 100 MeV) uncertainty of 2 to 3% is expected because of uncertainties in the pair production cross-section, particularly the 10 to 30 MeV photonuclear giant

**TABLE II**  
**Comparison of effective atomic numbers from the present study with experimental values.**

Description	Literature	Energy (keV)					
		279.1	661.6	662	1115.5	1173	1332
Steel 1	b	25.90	25.90	–	25.90	–	–
	c	25.29	25.24	25.24	25.23	25.23	25.22
Steel 2	b	25.80	26.00	–	26.00	–	–
	c	25.35	25.31	25.31	25.30	25.29	25.29
Steel 3	b	25.80	26.00	–	25.90	–	–
	c	24.92	24.83	24.83	24.81	24.81	24.81
Steel 4	b	25.90	25.90	–	25.90	–	–
	c	25.67	25.65	25.65	25.65	25.63	25.64
SS316L	a	–	–	26.61	–	26.85	26.88
	c	25.40	25.29	25.29	25.28	25.27	25.27
CS516	a	–	–	25.66	–	25.89	26.03
	c	25.41	25.37	25.37	25.36	25.35	25.36

a. Fakarudin *et al.*, 2011

b. Prasanna and Kumar, 2012

c. Present study

resonance region. Above 100 MeV the uncertainty is expected to be 1 to 2%. The uncertainty in mass attenuation coefficients is expected to more than double below 10 keV photon energy. The investigation values in our present study are very useful, with an acceptable range of uncertainties, as the steel will be used at photon energies of 5 keV and above. In the case of the Auto- $Z_{\text{eff}}$  software, the uncertainties in the effective atomic number are 1-2% at higher energies away from absorption edges.

#### 4. Conclusions

The present study provides the values of the gamma-ray mass attenuation coefficients, effective atomic numbers and electron densities of carbon steel and stainless steels. Our results show that the  $\mu/\rho$ ,  $Z_{\text{pleff}}$  and  $N_{\text{eff}}$  values of the alloys are photon energy- and composition element-dependent and vary mainly in the photo-electric absorption and Compton scattering regions. In the photo-electric absorption region low energy (<10 keV), very high values of  $Z_{\text{pleff}}$  are observed. These high values of  $Z_{\text{pleff}}$  require further experimental investigations.

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## PHOTON INTERACTION WITH CARBON STEEL AND STAINLESS STEELS

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