Radionuclide contents and radiological risk to the population due to raw minerals and soil samples from the mining sites of quality ceramic and pottery industries in Akwa Ibom, Nigeria

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ABSTRACT Samples of domestically produced industrial raw minerals and soil samples from three mining sites of quality ceramic/smelting and pottery industries in Akwa Ibom, Nigeria, were collected and analyzed for their 226Ra, 232Th and 40K contents using gamma-ray spectroscopy. The range of activity concentrations of the radionuclides in the industrial raw minerals were 17.55 ± 1.63 to 80.99 ± 2.61 Bq.kg⁻¹ for 226Ra, 7.64 ± 0.77 to 23.94 ± 0.92 Bq.kg⁻¹ for 232Th and 63.22 ± 3.43 to 503.90 ± 5.69 Bq.kg⁻¹ for 40K, while in the soil samples they varied from 2.87 to 34.78 Bq.kg⁻¹, 7.02 to 24.47 Bq.kg⁻¹ and 7.05 to 162.81 Bq.kg⁻¹ for 226Ra, 232Th and 40K, respectively. These results, along with the estimated absorbed dose rates, annual effective dose rates, radium equivalent (Raeq), external hazard index (Hex), internal hazard index (Hin) and representative of the gamma index (Iγ) are presented. The results obtained were below the internationally accepted safe limits. Therefore, the analyzed samples could be used in the local industries in the area as component raw materials and/or as building materials. Also, the mining activities of these minerals in the area have not significantly affected the natural radiation dose levels in the area, hence the resulting dose to the population is therefore considered generally low.

Keyword: Natural radionuclides / industrial raw minerals / radiological risk / population / Nigeria

RÉSUMÉ Contenu en radionucléides et risque radiologique aux populations d’échantillons de sols et de minéraux bruts issus de sites miniers des industries de la poterie et de la céramique de qualité à Akwa Ibom, Nigeria.

Des échantillons de minéraux bruts industriels produits domestiquement ainsi que de sols issus de trois sites miniers des industries de céramique de qualité et de Poterie situés à Akwa Ibom, Nigeria ont été collectés et analysés pour leur contenu en 226Ra, 232Th et 40K par spectrométrie gamma. Le niveau d’activité en radionucléides dans les minéraux bruts industriels variaient de 17,55 ± 1,63 à 80,99 ± 2,61 Bq.kg⁻¹ pour le 226Ra, de 7,64 ± 0,77 à 23,94 ± 0,92 Bq.kg⁻¹ pour le 232Th et de 63,22 ± 3,43 à 503,90 ± 5,69 Bq.kg⁻¹ pour le 40K alors que dans les échantillons de sol, ce niveau variait de 2,87 à 34,78 Bq.kg⁻¹, de 7,02 à 24,47 Bq.kg⁻¹ et de 7,05 à 162,81 Bq.kg⁻¹ respectivement pour le 226Ra, le 232Th et le 40K. Ces résultats sont présentés ainsi que les estimations des débits de dose absorbée, les débits de dose effective annuelle, la dose équivalente en radium (Raeq), l’indice de détriment externe (Hex), l’indice de...
détriment interne ($H_d$) et une représentation de l’indice d’exposition aux radiations gamma ($I_{\gamma}$). Les résultats obtenus sont inférieurs aux limites de sûreté internationalement acceptées. De ce fait, les échantillons analysés peuvent être utilisés par l’industrie locale comme matières premières et/ou comme matériaux de construction. De plus, les activités minières locales produisant ces minéraux n’affectent pas de façon significative les niveaux de dose de radiation naturelle dans la région, et par conséquent la dose résultante à la population est ainsi considérée comme généralement basse.

1. Introduction

The radionuclides in mineral soil generate a significant component of the background radiation exposure to the population (Goddard, 2002). The environmental effect of radiation from mineral resources depends on the use of the raw material or end product of such minerals in the environment. Kaolin and the clay mineral kaolinite are natural components of the soil and occur widely in ambient air as floating dust. Kaolinite is formed mainly by decomposition of feldspars (potassium feldspars), granite and aluminum silicates. It is also not uncommon to find kaolin deposited together with other minerals (illite, bentonite) (Grim, 1968). Clay is a widely distributed, abundant mineral resource of major industrial importance for an enormous variety of uses (Ampian, 1985). Illite is widely distributed in nature, abundant, and often the dominant clay mineral in soil, terrestrial deposits, sedimentary rocks, freshwater sediments, and most deep-sea clays (Grim, 1968). Kaolin and common clay may contain varying amounts of silica; a short summary is presented in recent assessments of quartz (IARC, 1997a, b).

Owing to inherently complex physical, chemical and mineralogical characteristics, clays are used mainly in the manufacture of bricks, portland and other cements, concrete blocks and structural concrete, and refractory and ceramics (electrical porcelain, fine china and dinnerware, floor and wall tiles, and pottery). These minerals, especially clay, are used in construction of dwellings by local residents in Akwa Ibom State of Nigeria.

Due to the health risks associated with the exposure to indoor radiation, many governmental and international bodies such as the International Commission on Radiological Protection (ICRP), the World Health Organization (WHO), etc. have adopted strong measures aimed at minimizing such exposures. This is imperative because the most significant exposure as regards the radiation health burden is due to the isotopes of radon ($^{222}$Rn and $^{220}$Rn). They are the decay products of radium ($^{226}$Ra and $^{224}$Ra) and are the members of the decay series of $^{238}$U and $^{232}$Th, respectively. Radon and its short-lived daughters are alpha emitters. They consequently become a major source of internal exposure of the respiratory tracts.
when inhaled (IPCS, 2000; Stoulos et al., 2003; Ferdoes et al., 2007), hence the call for the measurement and assessment of natural radionuclides in industrial raw minerals and construction materials worldwide (Stoulos et al., 2003; Turhan Seref, 2009; Ferdoes et al., 2007; Chang et al., 2008; Abel-Ghany et al., 2009).

In Nigeria there is a concerted effort towards determining the radionuclide concentration levels in different raw mineral and building materials (Farai and Ademola, 2001; Ademola, 2005; Ademola et al., 2008; Obed et al., 2005) and industrial wastes and by-products from some industries (Jibiri and Adewuyi, 2008; Farai et al., 2009). It has not been possible to carry out a nationwide survey of these materials because of logistics and financial set-backs. However, most radioactivity studies in building and raw materials have remained localized and it is envisaged that a compilation of these results from different regions will generate the overall requisite information on building materials with a potential radiological impact on human population in Nigeria.

This present study was designed to add to and enhance the existing information on radioactivity contents from different building materials in Nigeria with particular interest in Akwa Ibom State in the South-South region of the country. Presently, there is no data existing on the concentration of these natural radionuclides in minerals used for industrial raw minerals and construction purposes in the state, especially for the industrial raw minerals considered in this study.

The knowledge of radiation levels in the environment is imperative; hence this study is expected to yield data that will provide information that may be used to assess the health effects on the population due to the usage of the final products from these minerals, and also terrestrial radiation exposure around these mining sites. The objectives of the present study are to: (i) determine the activity concentrations of $^{226}\text{Ra}$, $^{232}\text{Th}$ and $^{40}\text{K}$ in industrial raw minerals and soil samples at the mining sites of industries in Akwa Ibom; (ii) determine radiological hazard indices such as radium equivalent activity ($R_{eq}$), absorbed gamma dose rate ($D$), the external ($H_{ext}$), internal ($H_{in}$) and the representative gamma ($I_{\gamma}$), (iii) provide radiometric data that may be useful in future epidemiological studies, and (iv) provide information that will enhance the radioactivity data of the Nigerian environment.

2. Materials and methods

2.1. Sample collection

Akwa Ibom State lies between latitudes $4^\circ32'$ and $5^\circ33'$ North; and longitudes $7^\circ25'$ and $8^\circ25'$ East. The landscape of Akwa Ibom is mostly flat. This is because the underlying geology of the state is predominantly coastal plain sediments. The
coastal nature of the state makes it the natural deposit of a mosaic of marine, deltaic, estuarine, lagoonal and fluviolacustrine material. This favors the large expanse of deposits of mineral sources such as kaolin, kalonite, illite, etc. in the area, and extensive mining activities. Pottery, quality ceramics and smelting industries amongst others are widespread across the state. There are three major mining sites in the state; kaolin, laterite and silica sand, and they belong to the quality ceramics/smelting and pottery industries.

About six domestically produced industrial raw minerals, and fifteen samples of soil at each of the three mining sites were collected and also from the immediate communities at relatively different distances from these mining sites. The domestic minerals were sampled directly by visiting the mine sites. These are samples that may likely find their way into the environment and consequently contaminate it. This was considered important so that spatial radionuclide concentration levels could be obtained for the assessment of the likely operational effects of the activities of the industries on the immediate environment and the population. The soil samples and the raw materials were collected into labeled polythene bags and were transferred to the Radiation and Health Physics Research Laboratory, University of Ibadan, for gamma-ray analysis. The sample types and the number of each type that was collected are presented in Table I.

### 2.2. Sample preparation

The collected samples were pulverized and sieved through a 0.2 mm mesh with a crusher and ceramic mortar, and then homogenized to obtain the same structural matrix after which, a mass of 200 g each was weighed using a Scout Pro weight machine by Ohaus with precision of 0.1 g. The powdered samples were transferred to cylindrical screw cap plastic containers of dimensions 6.8 cm × 7.3 cm and

<table>
<thead>
<tr>
<th>Sample code</th>
<th>Name of raw mineral</th>
<th>No. of mineral samples</th>
<th>No. of soil samples</th>
<th>Industries</th>
</tr>
</thead>
<tbody>
<tr>
<td>Rm1</td>
<td>Kyanite</td>
<td>1</td>
<td>-</td>
<td>QC*</td>
</tr>
<tr>
<td>Rm2</td>
<td>Laterite</td>
<td>1</td>
<td>15</td>
<td>QC*</td>
</tr>
<tr>
<td>Rm3</td>
<td>Kaoline</td>
<td>1</td>
<td>15</td>
<td>QC*</td>
</tr>
<tr>
<td>Rm4</td>
<td>Kaolinitic clay</td>
<td>1</td>
<td>-</td>
<td>Pot. Ind.</td>
</tr>
<tr>
<td>Rm5</td>
<td>Illite (Common clay)</td>
<td>1</td>
<td>-</td>
<td>Pot. Ind.</td>
</tr>
<tr>
<td>Rm6</td>
<td>Silica sand</td>
<td>1</td>
<td>15</td>
<td>QC*</td>
</tr>
</tbody>
</table>

QC* = Quality Ceramic Industry (Industrie de la céramique de qualité), Itu. Pot. Ind. = Pottery Industry (Poterie), Ikot Eboom.
sealed tightly with covers using masking/transparent tape to prevent the escape of 222Rn, the daughter of 226Ra. The sealed samples were weighed and kept for more than 5 weeks so as to re-establish secular equilibrium between radium isotopes and their progenies following sample collection and preparation procedures.

The detector used for the radioactivity measurements was a 10 cm lead-shielded 76 mm × 76 mm NaI(Tl) detector crystal (802 series Canberra Inc.) coupled to a Canberra Series 10 plus Multichannel Analyzer (MCA – Model: 1105) through a preamplifier. It has a resolution (full weight at half maximum – FWHM) of about 8% at energy of 0.662 MeV for 137Cs which is considered adequate to distinguish the gamma-ray energies of interest in this study. The choice of radionuclide to be detected was predicated on the fact that the NaI(Tl) detector used in this study has high efficiency.

2.3. Energy calibration

The purpose of calibration is to determine, under a controlled set of standard conditions, the indication of an instrument as a function of the value of the measured (the quantity intended to be measured) and this was carried out over the complete range of indications of the system. The calibration of the detector to determine the equation relating energy to channel number was carried out by placing different gamma sources of known energies from Nucleus Inc., Oak Ridge, USA, of IAEA EMS calibrated gamma sources and Geological Certified Reference Material from Radiometric Measurement from the International Atomic Energy Agency (IAEA), Vienna. The gamma sources were placed on the detector at a distance 7 cm from the surface of the detector. After a preset counting time of 100 s, the channels of the various photopeaks corresponding to the gamma energies were identified. This was done for the purpose of identifying the various radionuclides that may be present in the sample through the energies they emit.

2.4. Detection efficiency calibration

The second stage in the calibration process was the determination of the gamma-ray counting efficiencies over the energy range 0.511–2.615 MeV. This was done by converting the count per second (cps) under the photopeaks to the activity concentration (Bq.kg⁻¹) of certified reference standard samples. The standard gamma source was prepared by Rocketdyne Laboratories, Canoga Park, California, USA. This is traceable to a mixed standard gamma source (ENV94084 – 200 g) by Analytics Inc., Atlanta, Georgia (USA). The certified standard sources have activity concentrations of 479.15 Bq.kg⁻¹ ± 4.89%, 566.47 Bq.kg⁻¹ ± 1.47% and 11.60 Bq.kg⁻¹ ± 6.61% for 40K, 226Ra and 232Th, respectively. The reference standard source was placed symmetrically on top of the detector and counted for
36 000 s (10 h) after which the counting efficiencies \( E_p \) at different gamma-energies were determined using equation (1) (Farai and Ademola, 2001; Ademola, 2005):

\[
E_p = \frac{A_{net}}{T_s C \gamma M_s}
\]

where \( A_{net} \) is the net area under the photopeak of each radionuclide, \( C \) is the activity concentration of each radionuclide in the standard reference source, \( T_s \) is the counting time in s, \( \gamma \) is the gamma yield (a fraction of the gamma rays of the particular energy per disintegration) and \( M_s \) is the mass of the sample.

### 3. Results and discussion

#### 3.1. Activity concentrations of the industrial raw minerals and soil samples

From the gamma spectrometric analysis, three naturally occurring radionuclides were detected and measured. The obtained values of the measured activity concentrations of \(^{226}\text{Ra} \), \(^{232}\text{Th} \) and \(^{40}\text{K} \) in the industrial raw mineral samples are presented in Table II. As can be seen from the table, the concentrations varied for \(^{226}\text{Ra} \) (17.55 ± 1.63–80.99 ± 2.61) Bq.kg\(^{-1}\), \(^{232}\text{Th} \) (7.64 ± 0.77–23.94 ± 0.92) Bq.kg\(^{-1}\) and \(^{40}\text{K} \) (63.22 ± 3.43–503.90 ± 5.69) Bq.kg\(^{-1}\). Also, \(^{40}\text{K} \) and \(^{232}\text{Th} \) were found to be high in illite clay (Rm 5) with activity concentrations of 503.92 ± 5.69 Bq.kg\(^{-1}\) and 23.94 ± 0.92 Bq.kg\(^{-1}\), respectively, while kaolin (Rm 3) had the
The highest measured value of $^{226}$Ra, with an activity concentration of $80.99 \pm 2.60$ Bq kg$^{-1}$. The activity concentrations of the radionuclides in this study are comparable with the values in concrete blocks in Ibadan, Nigeria (Farai and Ademola, 2001), but lower than those determined for granite samples in Ondo State (Ademola and Ayeni, 2010).

Similarly, from the measured gamma-ray spectra of $^{226}$Ra, $^{232}$Th and $^{40}$K, activity concentrations of the soil samples were determined for the three mining sites (A, B and C). The range and mean of the activity concentrations are presented in Table III. From the results obtained, $^{226}$Ra and $^{232}$Th have the highest activity concentrations of $34.78 \pm 1.95$ and $21.56 \pm 0.93$ Bq kg$^{-1}$, respectively, which are high when compared with the values obtained by Ademola et al. (2008) in Igbeti, Nigeria, which has a similar sedimentary formation to the study area region. The variation and similarities are in agreement with previous studies that terrestrial radioactivity to a large extent depends on the local geology and other related factors in the environment (Wollenberg and Smith, 1990). According to Orabi et al. (2006) and Al-Jundia et al. (2003), the concentrations of these radionuclides in soil are determined by the radioactivity of the rock and also the nature of the process of the formation of the soils.

### 3.2. Determination of radiation hazard indices

#### 3.2.1. Assessment of radium equivalent activity in the industrial raw minerals and soil samples

To represent the activity levels of $^{226}$Ra, $^{232}$Th and $^{40}$K by a single quantity, which takes into account the radiation hazards associated with them, a common
A radiological index has been introduced. This index is called radium equivalent (Ra$_{eq}$) activity and is expressed mathematically by UNSCEAR (2000):

$$Ra_{eq} (Bq.kg^{-1}) = A_{Ra} + 1.43 A_{Th} + 0.077 A_{K}$$  \hspace{1cm} (2)

where $A_{Ra}$, $A_{Th}$ and $A_{K}$ are the activity concentrations of 226Ra, 232Th and 40K, respectively.

It may be noted that 238U was replaced with the decay product 226Ra because there may be disequilibrium between 238U and 226Ra. The maximum permissible value of the radium equivalent activity is 370 Bq.kg$^{-1}$ (UNSCEAR, 1982): this evaluation will assist in determining if soil from the study area could be used for building of residential houses. In the above relation (Eq. (2)), it has been assumed that 10 Bq.kg$^{-1}$ of 226Ra, 7 Bq.kg$^{-1}$ of 232Th and 130 Bq.kg$^{-1}$ of 40K produce an equal gamma dose (Beretka and Mathew, 1985). The results of the radium equivalent obtained are presented in Table IV. Similarly, from equation (2) the radium equivalent activity of the collected soil samples for the three sites were obtained. These are presented in Table V. The values varied as follows; for Site A (kaolin) (30.47–50.15 Bq.kg$^{-1}$) with a mean value of 40.01 ± 1.22 Bq.kg$^{-1}$; Site B (laterite) (38.78–64.69 Bq.kg$^{-1}$) with a mean value of 49.67 ± 3.51 Bq.kg$^{-1}$ and Site C (silica sand) (25.15–68.83 Bq.kg$^{-1}$) with a mean value of 46.66 ± 2.83 Bq.kg$^{-1}$. These values are below the maximum permissible value of 370 Bq.kg$^{-1}$ (UNSCEAR, 2000).

### 3.2.2. Determination of gamma absorbed dose rate in air

The absorbed dose rates ($D$) due to gamma radiation in air at 1 m above the ground surface for the uniform distribution of the naturally occurring radionuclides
The absorbed dose rate in air for the collected soil samples was calculated for the three sites (A, B and C) using equation (3). The range and mean results are also presented in Table III. The highest average absorbed dose of 23.37 ± 3.79 nGy.h⁻¹ was obtained at site B (laterite) and the lowest average of 3.51 ± 0.25 nGy.h⁻¹ at site C (silica sand). All these values are lower than the world average outdoor terrestrial dose value of 55 nGy.h⁻¹ (UNSCEAR, 2000).

### 3.2.3. Assessment of annual effective dose rate in air

The absorbed gamma dose rates in air are more often related to the human gamma dose in order to assess the effectiveness of the gamma dose in causing damage to human tissues. Damage to the tissues may result in somatic and genetic health effects. To estimate the annual effective dose rates, the conversion coefficient from the absorbed dose in air to the effective dose (0.7 Sv.Gy⁻¹) and the outdoor occupancy factor (0.2) proposed by UNSCEAR (2000) were used. Therefore, the annual effective dose rate (mSv.y⁻¹) was calculated using the formula (Jibiri and Adewuyi, 2008; Jibiri and Emelue, 2008)

\[
E_{\text{air}} = T f Q D_{\text{air}} \xi 
\]

where \( E_{\text{air}} \) is the effective dose rate (mSv.y⁻¹), \( T \) is the time, \( f \) is the outdoor occupancy factor which corrects for the average time spent outdoors (0.2), \( Q \) is the

### Table V

The range and mean of the Radium equivalent dose \( R_{\text{eq}} \) (Bq.kg⁻¹), and external (\( H_x \)), internal (\( H_i \)) and gamma (\( I_{\text{\gamma r}} \)) radiation exposure of soil samples at the three sites.

<table>
<thead>
<tr>
<th>Sample locations</th>
<th>( R_{\text{eq}} ) (Bq.kg⁻¹)</th>
<th>( H_x )</th>
<th>( H_i )</th>
<th>( I_{\text{\gamma r}} )</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Range</td>
<td>Mean</td>
<td>Range</td>
<td>Mean</td>
</tr>
<tr>
<td>Site A</td>
<td>30.47–50.15</td>
<td>40.01 ± 4.88</td>
<td>0.11–0.20</td>
<td>0.21–0.37</td>
</tr>
<tr>
<td>Site B</td>
<td>38.78–64.69</td>
<td>49.67 ± 7.51</td>
<td>0.14–0.24</td>
<td>0.27–0.46</td>
</tr>
<tr>
<td>Site C</td>
<td>25.15–68.83</td>
<td>46.66 ± 11.36</td>
<td>0.18–0.28</td>
<td>0.33–0.48</td>
</tr>
</tbody>
</table>

\(^{226}\text{Ra}, \ ^{232}\text{Th} \text{ and } ^{40}\text{K}\) were calculated based on guidelines provided by UNSCEAR (2000). We assumed that the contributions from other naturally occurring radionuclides were insignificant. Hence, \( D \) can be calculated according to UNSCEAR (2000) as:

\[
D \text{ (nGy.h⁻¹)} = 0.462 A_{\text{Ra}} + 0.621 A_{\text{Th}} + 0.0417 A_{\text{K}}. \quad (3)
\]
quotient of the effective dose rate and absorbed dose rate in air (0.7 Sv Gy⁻¹), ξ is the factor converting nano (10⁻⁹) into milli (10⁻³); and \( D_{air} \) is the absorbed dose rate in air (nGy h⁻¹). Using the total absorbed dose rate values from equation (4), the baseline annual effective dose was calculated using the mean total absorbed dose in Table II. From the results obtained, the effective dose rate in air for the three sites (A, B and C) ranged from 0.0043–0.0293 mSv yr⁻¹ with a mean value of 0.02 mSv yr⁻¹ (20 \( \mu \)Sv yr⁻¹). These values were below the prescribed limits, since the acceptable dose equivalent for the public is 1 mSv yr⁻¹, or 20 mSv yr⁻¹ for radiation workers (ICRP, 1991).

### 3.2.4. External hazard index (Hₑₓ)

An extensively used hazard index (reflecting the external exposure) called the external hazard index \( Hₑₓ \) is defined by (UNSCEAR, 2000):

\[
Hₑₓ = \frac{A_{Ra}}{370} + \frac{A_{Th}}{259} + \frac{A_{k}}{4810}.
\]

The external index for the industrial raw minerals was calculated using equation (5); the results are presented in Table IV. The values varied from 0.07 to 0.36 with a mean value of 0.22. Similarly, the external hazard index of the collected soil samples was also calculated using equation (5) and the results are presented in Table V. The values obtained ranged for Site A from (0.01–0.14); Site B (0.11–0.18) and Site C (0.09–0.19).

### 3.2.5. Internal hazard index (Hᵢᵣ) for the raw minerals

In addition to the external hazard index, radon and its short-lived products are also hazardous to the respiratory organs. Internal exposure to radon and its daughter products is quantified by the internal hazard index \( Hᵢᵣ \) which is given by Beretka and Mathew (1985) in equation (6):

\[
Hᵢᵣ = \frac{A_{Ra}}{185} + \frac{A_{Th}}{259} + \frac{A_{k}}{4810}.
\]

The values of the indices \( (Hₑₓ, Hᵢᵣ) \) must be less than unity for the radiation hazard to be negligible (Beretka and Mathew, 1985). The internal index \( (Hₑₓ) \) for the industrial raw minerals was calculated using equation (6); the values varied from 0.15 to 0.57 with a mean value of 0.37. These values are also presented in Table IV.

Similarly, the internal index of the collected soil samples was also calculated for the three sites (A, B and C). The results obtained are presented in Table V. The
values ranged for site A from (0.11–0.20); site B (0.14–0.24) and site C (0.12–0.28). It is clear that the results obtained were below the criterion of unity (UNSCEAR, 2000).

3.2.6. Gamma index ($I_\gamma$) for the raw mineral samples

According to the European Commission (1999), the following gamma activity concentration index ($I_\gamma$) (representative level index) is derived for identifying whether a dose criterion is met:

$$I_\gamma = \frac{A_{Ra}}{150} + \frac{A_{Th}}{100} + \frac{A_k}{1500}$$

(7)

The index $I_\gamma$ is correlated with the annual dose due to the excess external gamma radiation caused by superficial material. Values of index $I \leq 1$ correspond to 0.3 mSv/yr$^{-1}$, while values of $I \geq 3$ correspond to 1 mSv/yr$^{-1}$ (EC, 1999). Gamma index ($I_\gamma$) radiation exposure was calculated from equation (7): the results are presented in Table IV. The results ranged from 0.27 to 0.96 with a mean value of 0.61. These values are below the criterion of unity corresponding to an annual effective dose of 0.3 mSv (EC, 1999).

Again, the gamma index of the collected soil samples for the three sites was calculated and the results obtained are also presented in Table V. The values, as can be seen from the table, ranged for site A from (0.21–0.37); site B (0.27–0.46) and site C (0.18–0.48). The range of values for the external, internal and gamma indices obtained for the soil samples were below the criterion of unity corresponding to an annual effective dose of 0.3 mSv yr$^{-1}$ (EC, 1999). Hence, the use of soil samples as building material is seen not to pose any significant radiation hazard to the population in the study area.

4. Conclusion

In view of worldwide concern about the radionuclide contents of various industrial raw materials, radiological assessment of the activity concentrations of different industrial raw minerals (kaolin, illite, kyanite, laterite and silica sand) in Akwa Ibom State, South-South Nigeria, were carried out. From the results obtained, the total absorbed dose rate in air due to soil samples for the three sites ranged from 1.69 to 30.00 nGy.h$^{-1}$ with a mean of 18.14 ± 2.19 nGy.h$^{-1}$. This corresponds to an annual effective dose range of 0.0043–0.0293 mSv.yr$^{-1}$. These values and other radiological indices such as the radium equivalent, and internal, external and representative gamma indices estimated for the soil and raw mineral samples were found to be below the prescribed safe limits. Therefore, use of these samples in the
local industries as component raw materials or as building materials for dwellings is seen not to pose any significant radiation health hazard to the population in the study area. Also, the mining activities of these minerals in the area have not significantly affected the natural radiation dose levels in the area.

REFERENCES


RADIONUCLIDE CONTENTS AND RADIOLOGICAL RISK TO THE POPULATION


